



Research Article

EFFECT OF ZINC OXIDE NANOPARTICLES ON *CANDIDA ALBICANS* OF HUMAN SALIVA (*IN VITRO* STUDY)

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ABSTRACT

The potential use of zinc oxide and other metal oxide nanoparticles in biomedical are gaining interest in the scientific and medical communities, largely due to the physical and chemical properties of these nanoparticles, therefore there is an urgent need to develop new classes of antimicrobial agents, and recent studies demonstrate that hold a considerable promises. *Candida albicans* were isolated from saliva of forty eight volunteers of both sexes their age range between 18-22 years and then purified and diagnosed according to morphological characteristic and biochemical tests. Different concentrations of ZnO NPs were prepared from the stock solution; all the experiments were conducted in vitro. Disk diffusion method was used to study the sensitivity of *Candida albicans* to different concentrations of zinc oxide nanoparticles in comparison to effect of de-ionized water. *Candida albicans* were sensitive to all concentrations (0.01, 0.05, 0.1, 0.5, 1, 3 and 5.8) mg / ml of the zinc oxide nanoparticles solution in comparison to de-ionized water, revealing a highly significant difference in all concentrations.

This study revealed that zinc oxide nanoparticles were effective against *Candida albicans*.

Keywords: *Candida albicans*, ZnO NPs, Human saliva.

INTRODUCTION

Nanotechnology is referred to the emerging technology involving fabrication or application of nanosized structures or materials (1). Nanoparticles are commonly defined as particles with the size of at least one dimension ranging from 1 to 100 nm, which serve as a bridge between bulk materials and atoms/molecules (2). Zinc oxide nanoparticles (ZnO NPs), is an inorganic white powder insoluble in water (3). Zinc oxide nanoparticles have been shown to be useful antibacterial and antifungal agents when used as a surface coating on materials and textiles (4). ZnO nanorod arrays diminished the growth of *Candida albicans* with stable action for two months (5). *Candida albicans* is a commensal fungal species commonly colonizing human mucosal surfaces (6). *Candida* infections can be seen in various parts of the

body, including the skin, oral cavity, gastrointestinal tract and vagina. In the oral cavity *Candida* infection is sometimes visible as thrush, white/yellowish cream like patches on the oral mucosa and the tongue (7). The incidence of these organisms appears to increase with the age. These fungi can cause characteristic infections that may be quiet serious and even life threatening (8). Some studies have shown a significant association between *C. albicans* and dental caries in children and young adults (9). The findings from in vitro and animal studies attributing a role for *C. albicans* in caries development and/or progression were solidified by data from a clinical study where the occurrence of caries in children was positively correlated with the frequency of oral candidal carriage (10). ZnO NPs have shown to have a

photo-catalytic effect (11), which is related their effectiveness as inhibitor of bacteria and fungi (Figure 1).

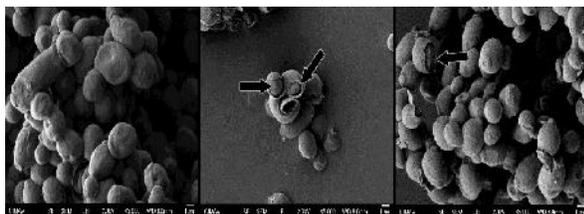


Figure 1: Cell wall damage of *C.albicans* exposed to nanoparticles. Left: Untreated control; center: AgNPs; right: ZnO NPs (12).

MATERIALS AND METHODS

Collection of Saliva Samples

Collection of stimulated salivary samples from forty eight apparently healthy college students from Baghdad University /Collage of Dentistry their age range between (18-22 years) of both sexes. The collection of stimulated salivary samples was performed under standard condition following instruction cited by Tenovou and Lagerlof (13). Each individual was asked to chew a piece of Arabic gum (0.5-0.7) gm for one minute, then remove all saliva by expectoration, after that chewing was continued for ten minutes with the same piece of gum and saliva collected in a sterile screw capped bottle. After disappearance of salivary foam, 0.1 ml of saliva is transferred to 0.9 ml of sterile phosphate buffer saline of pH 7.0-7.2 for microbiological analysis.

Preparation of Culture Media

Sabouraud Dextrose Agar (SDA)

The medium which is selective one for cultivation and isolation of *Candida albicans* was prepared and sterilized according to manufacturer's leftlet; 65gm were suspended in 1000 distilled water. Sterilization was done by autoclaving at 121°C at 15 pounds per square inch for 15 minutes, left to cool to 45-50 °C and then chloramphenicol antibiotic (2g for each 1000 ml of media) was added and poured into petri dishes, left to solidify then put them in incubator at 37°C for 24 hours then stored in refrigerator until being used.

Brain Heart Infusion Broth (BHI)

Preparation of the media was according to the manufacturer's instruction which involved the suspension of 37gm in one liter of de-ionized water. After being completely dissolved, the pH adjusted to (7.2) The media

was sterilized by autoclave at 121oC at 15 pounds per square inch for 15 minutes then left to cool down to room temperature and thereafter kept in the refrigerator until use.

Mueller Hinton Agar (MHA)

These were prepared according to manufacturer's instruction which involved the suspension of 38 gm in one liter of de-ionized water, after being completely dissolved with boiling, it was sterilized by autoclave, then left to cool at 45- 50oC, poured and left to solidify then put them in incubator at 37°C for 24 hours then stored in refrigerator until being used.

Isolation of *Candida albicans*

After mixing of saliva as mentioned previously, ten- folds dilution was performed, from dilution(10⁻¹, 10⁻²) of salivary samples then 0.1 ml was taken and spread on sabouraud dextrose agar (SDA), and the plates were incubated aerobically for 48hr at 37oC (14).

Identification of *Candida albicans*

A. Colony morphology

Colonies of *C.albicans* appeared smooth creamy in color with a yeast odor and typically medium size (1.5-2 mm) diameter which later developed into high convex, off-white larger colonies after about 2 days (15).

B. Gram stain

The same procedure as described for *Streptococcus mutans* was used, *C.albicans* appeared as Gram- positive small oval or budding yeast cell.

C. Germ tubes formation

Very small inoculums from isolated colonies were suspended in 0.5 ml of normal human serum. The inoculated tubes were incubated at 37°C for 3hr. After incubation, a drop of yeast suspension was placed on a clean microscopic slide covered with a cover slip and examined under low power magnification for presence of germ tube. Production of germ tubes is characteristic of *Candida albicans* (16).

D- Identification of *Candida albicans* by Rapid Yeast Plus System

Rapid Yeast Plus System has several reaction cavities molded into the periphery of plastic disposable tray. Reaction cavities contain dehydrated reactions and the tray allows the simultaneous inoculation of each cavity with a predetermined amount of inoculum (17).

Purification and Maintenance of *Candida albicans*

A single colony from *C. albicans* was transferred to 10 ml sterile brain heart infusion (BHI) broth and then incubated for 24 hours aerobically at 37°C. The purity of isolates was checked by re- inoculation of 0.1ml of culture broth BHI on SDA. The plates were incubated aerobically for 48hr at 37°C, then one colony from each isolates was transferred to 10 ml of sterile BHI broth and then incubated for 24 hours aerobically at 37°C (18).

Activation of *Candida albicans*

Inoculums of *C. albicans* were activated by the addition of 0.1ml of pure broth culture to 10ml of BHI broth followed by incubation for 18 hr. At 37°C(19).

Characterization of Zinc Oxide nanoparticles

Zinc oxide nanoparticles provided from ministry of sciences and technology, with the concentration 5.8 mg/ml for stock solution and the particles size >50 nm papered by sol gel method. We make different concentration from the stock solution by using dilution law ($N_1V_1 = N_2V_2$). To confirm the activity of zinc oxide nanoparticles solution we make the UV-Vis spectra of ZnO NPs which was shown in Figure (3). The absorption peak of the prepared ZnO NPs was found at around 400-500nm.



Figure 2: Zinc oxide nanoparticles solution (stock solution in concentration of 5.8 mg/ml).

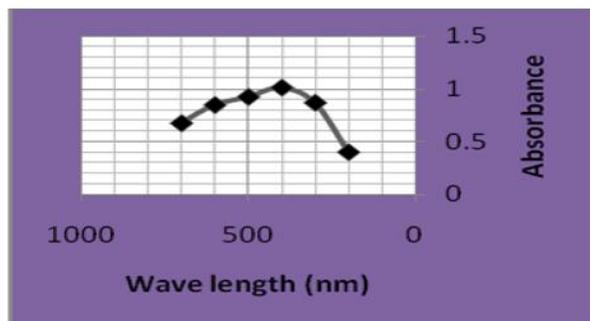


Figure 3: UV-Vis spectra of the ZnO NPs

Determining the Sensitivity of *Candida albicans* to Different Concentrations of ZnO NPs and de-ionized water

Fungal inoculums used was prepared by adding a few pure colonies of *Candida albicans* to 10 ml of sterile brain heart infusion broth (pH 7.0), then incubated aerobically for 18hr at 37°C. The disk diffusion method was used as antifungal susceptibility test. Disposable plates containing Muller-Hinton agar inoculated were applied to study the antibacterial effects of different concentrations of zinc oxide nanoparticles (0.01, 0.05, 0.1, 0.5, 1, 3, and 5.8) mg/ml compared with de-ionized water as negative control on Mueller Hinton Agar (MHA) media. These experiments were conducted on 48 isolates of *Candida albicans*.

The sensitivity of *Streptococcus mutans* and *Candida albicans* to ZnO NPs

1. A volume of 25 ml of MHA (pH 7) was poured into sterile petri dishes, left at room temperature for 24 hours.
2. To each plate 0.1 ml of activated *Candida albicans* inoculum was spread, left at room temperature for 20 minutes.
3. Eight filter papers (wattman no.1) of equal size (7 mm in diameter) were prepared; each filter was impregnated with 40 µl of ZnO NPs with different concentrations (0.01 , 0.05 , 0.1 , 0.5 , 1.0 , 3.0 , 5.8 mg/ml and de-ionized water) respectively, and put in each agar plate.
4. Plates were left at room temperature for 1 hour then incubated aerobically for *C. albicans* 24 hour at 37°C. Zone of inhibitions which appears as a clear zone of inhibition around disk were measured across the diameter of each filter paper by using a ruler, no inhibition zone indicated a complete resistance of *Candida albicans* to the agents.

RESULTS

Identification of *Candida albicans*

Colony morphology

Colony of *C. albicans* appeared smooth, creamy in color with yeast odor and typically medium sized (1.5-2mm) diameter within 2 days, they develop into high convex, off- white large colonies (Figure 4).

Microscopic examination

The slide was examined under light microscope, the rounded or oval yeast cells were Gram positive (Figure 5).

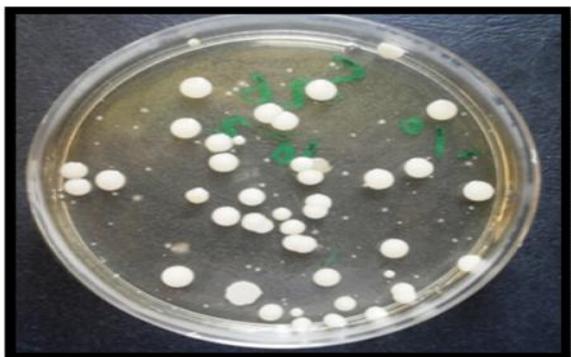


Figure 4: *C. albicans* colonies on SDA

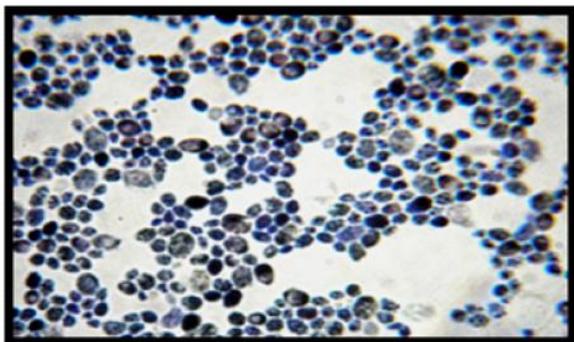


Figure 5: Gram's stain of *C. albicans*



Figure 6: Different shapes of germ tube of *C. albicans*



Figure 7: Biochemical identification of RapID Yeast Plus System for identification of *Candida albicans*

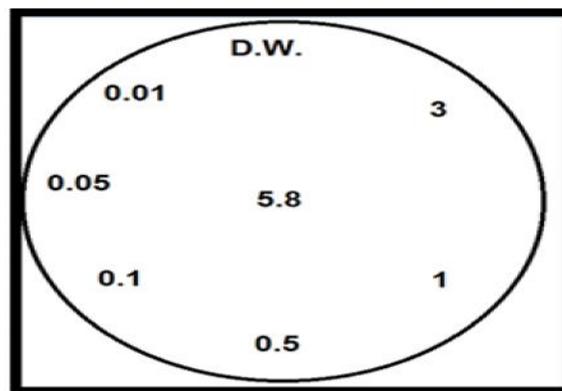


Figure 8: Sensitivity of *C. albicans* to different concentrations of ZnONPs

Germ Tube Formation

All isolates of *Candida albicans* under light microscope (100 x magnification) show the presence of germ tubes which is a characteristic feature of *C. albicans* (Fig. 6).

RapID Yeast Plus System for identification of *Candida albicans*:

The RapID Yeast Plus System Differential Chart illustrates the expected results for *Candida albicans*. Table (3-8) after incubation at 30°C in incubator for 4 hours (Figure 7).

Sensitivity of *Candida albicans* to Different Concentrations of ZnO NPs Solution and De-ionized Water

The diameter of inhibition zones for zinc oxide nanoparticles solution (clear zone of no growth for *Candida albicans* around each filter paper) were found to be increased as the concentration of the solution increased. The stock solution of ZnO NPs which equal 5.8 mg/ml showed higher zone of inhibition compared to other concentrations. De-ionized water showed no zone of inhibition (Figure 8).

Table 1: The Results of RapID Yeast Plus System for Identification of *Candida albicans*.

Test	Abbreviations	Positive result of <i>C.albicans</i>
Glucose	GLU	+
Maltose	MAL	+
Sucrose	SUC	-
Trehalose	TRE	-
Raffinose	RAF	-
Fatty acid ester	LIP	-
-Nitrophenyle-N-acetyle- ,D-galactosaminide	NAGA	+
-Nitrophenyl- , D-glucoside	GLU	+
-Nitrophenyl- , D-glucoside	GLU	-
-Nitrophenyl- , D-galactoside	ONPG	-
-Nitrophenyl- ,D- galactoside	GAL	-
-Nitrophenyl- , D-fucoside	FUC	-
-Nitrophenyl phosphate	PHS	V
-Nitrophenyl phosphorylcholine	PCHO	-
Urea	URE	-
Proline- -naphthylamide	PRO	+
Histidine - naphthylamide	HiST	V
Leucyl-glycine naphthylamide	LGY	V

Table 2: Test of Normality

Kolmogorov-Smirnov Test		
	Groups	Sig.
I.Zones	0.01	.000
	0.05	.000
	0.1	.001
	0.5	.005
	1	.000
	3	.000
	5.8	.020

Table 3: Descriptive statistics of inhibition zones of ZnO NPs and De ionized water on *C.albicans*

Groups	No.	Median	Mean	SD	Interquartile range
0.01 mg/ml	48	8.00	8.00	0.97	2.00
0.05 mg/ml	48	9.00	8.88	1.48	3.00
0.1 mg/ml	48	10.00	10.25	1.87	4.00
0.5 mg/ml	48	12.00	12.19	2.16	4.00
1 mg/ml	48	14.00	14.69	2.59	3.00
3 mg/ml	48	17.00	19.08	3.65	4.75
5.8 mg/ml	48	25.00	25.69	3.54	4.75
D.W	48	7.00	7.00	0.00	0.00

Table 4: Zones of inhibition of *C.albicans* to different concentrations of ZnO NPs.

	Groups	N	Mean Rank	Kruskal-Wallis H Test
I.Zones	1	48	88.53	Chi-Square = 334.877 df= 7 P 0.01 HS
	2	48	116.74	
	3	48	161.91	
	4	48	210.44	
	5	48	257.63	
	6	48	309.09	
	7	48	354.67	
	8	48	41.00	

Table 5: Mann-whitney U test of inhibition zones of *C. albicans*

Groups	Median	Mean Rank	U Value	Z Value	P Value	Sig.
0.01	8.00	39.67	728.00	3.22	0.01	HS
0.05	9.00	57.33				
0.01	8.00	32.41	379.50	5.80	0.01	HS
0.1	10.00	64.59				
0.01	8.00	25.36	41.50	8.22	0.01	HS
0.5	12.00	71.64				
0.01	8.00	24.59	4.50	8.49	0.01	HS
1	14.00	72.41				
0.01	8.00	24.50	0.00	8.53	0.01	HS
3	17.00	72.50				
0.01	8.00	24.50	0.00	8.51	0.01	HS
5.8	25.00	72.50				
0.05	9.00	38.23	659.00	3.67	0.01	HS
0.1	10.00	58.77				
0.05	9.00	29.32	231.50	6.85	0.01	HS
0.5	12.00	67.68				
0.05	9.00	24.85	17.00	8.38	0.01	HS
1	14.00	72.15				
0.05	9.00	24.50	0.00	8.51	0.01	HS
3	17.00	72.50				
0.05	9.00	24.50	0.00	8.50	0.01	HS
5.8	25.00	72.50				
0.1	10.00	37.48	623.00	3.93	0.01	HS
0.5	12.00	59.52				
0.1	10.00	27.56	147.00	7.44	0.01	HS
1	14.00	69.44				
0.1	10.00	24.50	0.00	8.49	0.01	HS
3	17.00	72.50				
0.1	10.00	24.50	0.00	8.47	0.01	HS
5.8	25.00	72.50				
0.5	12.00	35.79	542.00	4.52	0.01	HS
1	14.00	61.21				
0.5	12.00	25.80	62.50	8.03	0.01	HS
3	17.00	71.20				
0.5	12.00	24.51	0.50	8.46	0.01	HS
5.8	25.00	72.49				
1	14.00	32.07	363.50	5.81	0.01	HS
3	17.00	64.93				
1	14.00	24.85	17.00	8.35	0.01	HS
5.8	25.00	72.15				
3	17.00	29.97	262.50	6.56	0.01	HS
5.8	25.00	67.03				

Table 6: Mann-whitney U test of inhibition zones of *C. albicans*

Groups	Median	Mean Rank	U Value	Z Value	P Value	Sig.
0.01	8.00	64.50	384.00	6.76	0.01	HS
D.W	7.00	32.50				
0.05	9.00	65.00	360.00	6.88	0.01	HS
D.W	7.00	32.00				
0.1	10.00	71.50	48.00	8.76	0.01	HS
D.W	7.00	25.50				
0.5	12.00	72.50	0.00	9.04	0.01	HS
D.W	7.00	24.50				
1	14.00	72.50	0.00	9.05	0.01	HS
D.W	7.00	24.50				
3	17.00	72.50	0.00	9.06	0.01	HS
D.W	7.00	24.50				
5.8	25.00	72.50	0.00	9.04	0.01	HS
D.W	7.00	24.50				

Statistical Analysis Tests for Inhibition Zone of ZnO NPs on *Candida albicans*

All data for the inhibition zone for all groups concentrations of ZnO NPs (except 5.8) is not normally distributed (sig. 0.05) by Kolmogorov-Smirnov test for test of normality (Table 2). Therefore the tests used are non-parametric.

Descriptive statistics for inhibition zones are used to examine the differences among different concentrations of ZnO NPs (0.01, 0.05, 0.1, 0.5, 1, 3, 5.8) mg/ml with de-ionized water to make eight groups respectively in (Table 3) each group consist of 48 tests for the same concentration and the mean of inhibition zones measured in mm which included the diameter of filter paper in measurements, So when there is no inhibition zone we measure the diameter of filter paper only which is equal to 7 mm and refer to it in the (Table 4).

The Kruskal-Wallis H and Mann-Whitney U non-parametric statistical tests test are used to analyze (concentrations of ZnO NPs) groups: Results of Kruskal-Wallis test showed highly significant differences among all groups p 0.01 (Table 4).

Further analysis using a Mann-Whitney U Test was done to determine which of the eight groups of ZnO NPs was different from the other groups. The results of Mann-whitney U test with each other groups showed highly significant P 0.01 between all the groups (Table 5).

The results of Mann-whitney U test for the de-ionized water with other groups showed highly significant differences

between all groups (0.01, 0.05, 0.1, 0.5, 1, 3 and 5.8) P 0.01 (Table 6).

DISCUSSION

The results of present study showed the significantly antifungal activity against *Candida albicans* using the ZnO NPs as low as 0.01 mg/ml. As the concentration of ZnO NPs increased from 0.01 to 5.8 mg/ml the efficacy of ZnO NPs treatment was enhanced. Fungicidal activity of ZnO NPs was due to destroying cell membrane integrity (20). The results of the quantitative antifungal assessment by disk diffusion method are reported in Table (3) from which it is observed that the size of the inhibition zone (the antifungal activity) was found to depend strongly on the concentration of ZnO NPs, and these results agree with Eman et al in 2013 (21) who showed the fungicidal effect of ZnO NPs is concentration dependent and also indicate that the mechanism of the fungicidal action of ZnO NPs involves disrupting the membrane. These results agree with those obtained by Shi et al in 2010 and Lipovsky et al 2011(22,23), who recorded the ability of ZnO NPs to affect the viability of the pathogenic yeast, *Candida albicans*, as well as a concentration-dependent effect, but does not agree with present results in that the minimal fungicidal concentration of ZnO NPs was found to be (0.1 mg/ml). This concentration caused an inhibition of over 95% in the growth of *C. albicans*. While the current results observed that the minimal fungicidal concentration of ZnO NPs was found to be

(0.01 mg/ml). This may be due to the incubation time which was 5 days in study of Eman et al (2013) (21) while in our study it was 24 hours only, or may be due to *Candida albicans* that is isolated from the skin in study of Eman et al (2013) (21) which differ from *Candida albicans* isolate that is isolated from human saliva in that study.

The current study found that the least inhibition zone of *Candida albicans* was 8 mm in concentration (0.01 mg/ml) of ZnO NPs but this result in contrast to the results of Jehad et al (2012) (24) who showed that the best inhibition zone of *Candida albicans* was 18 mm in concentration 10 µg/ml which is equal to (0.01 mg/ml) the least conc. That is used in this study. This may be attributed to the antibacterial technique that used in both study as Jehad et al (2012) (24) used Agar Diffusion Technique in which hollows of 10 millimeters diameter wells were cut from the agar, and 0.1 ml of each of the tested solutions were poured into the wells. While in this study the disk diffusion method was used in which the filter paper of 7mm in diameter was impregnated in 40 µl only the difference in amount of ZnO NPs may cause the difference in inhibition zone diameter for the same concentration.

The study of (22) mentioned that, for almost all fungi, the central core of the cell wall is a branched -1, 3, 1, 6 glucan that is linked to chitin via a -1, 4 linkages. The binding of the oxides particles on the fungal cell surface through electrostatic interactions could be a possible mechanism.

CONCLUSIONS

- 1- Zinc oxide nanoparticles have inhibition effect in different concentrations on *Candida albicans*, starting from the concentration 0.05 mg/ml.
- 2- Sensitivity of *Candida albicans*, to ZnO NPs increase with the increase of concentration of ZnO NPs solution in comparison to de-ionized water.

REFERENCES

1. Maynard, Andrew D, Robert J. and Aitken. (2006): "Safe handling of nanotechnology." *Nature* 444(7117): 267-269.
2. Nel A, Xia T, Madler L, and Li N. (2006): "Toxic potential of materials at the nanolevel." *Science* 311(5761): 622-627.
3. Takahashi Y, Yoshikawa A, and Sandhu A. (2007): *Wide bandgap semiconductors: fundamental properties and modern photonic and electronic devices*. Springer; p. 357.

4. Abramov OV, Gedanken A, Koltyen Y, Perkas N, Perelshtein I, Joyce E and Mason TJ. (2009): Pilot Scale Sonochemical Coating of Nanoparticles Onto Textiles to Produce Biocidal Fabrics. *In Surface and Coatings Technology* 204: 718-722.
5. Eskandari M, Haghghi N, Ahmadi V, Haghghi F and Mohammadi SH (2011): Growth and investigation of antifungal properties of ZnO nanorod arrays on the glass. *Physica B* 406:112-4.
6. Calderone RA, (2012): *Candida and Candidiasis*. Washington: ASM Press.
7. Gow N AR and Gadd GM. (1994): *The Growing Fungus*. London: Chapman and Hall 324-32.
8. Shepherd M. (1986): The pathogenesis and host defense mechanisms of oral candidosis. *New Zealand dental journal* 82: 78-2.
9. de Carvalho FG, Silva DS, Hebling J, Spolidorio LC, and Spolidorio DM. (2006): Presence of mutans streptococci and *Candida* spp. in dental plaque/dentine of carious teeth and early childhood caries. *Archives of Oral Biology Journal* 51(11): 1024-8.
10. Raja M, Hannan A, and Ali K (2010): Association of oral candidal carriage with dental caries in children. *Caries Research* 44: 272-276.
11. He L, Liu Y, Mustapha A, and Lin M (2011) : Antifungal activity of zinc oxide nanoparticles against *Botrytis cinerea* and *Penicillium expansum*. *Microbiol Res* 166: 207-15.
12. Arzate-Quintana, Carlos, Sánchez-Ramírez, Blanca, Infante-Ramírez, Rocío, Piñón-Castillo Hilda Amelia], Montes-Fonseca, Silvia Lorena, Duarte- Moller, Alberto, Luna-Velasco, Antonia, and Orrantia-Borunda, Erasmo. (2013): An approach to the mechanism of the cytotoxic effect of Silver and Zinc Oxide Nanoparticles.
13. Tenovou J, Lagerlof F. and Saliva. In: Thylstrup A, Fejerskov O. (1996): *Textbook of clinical cariology*. 2nd ed. Munksgaard, Copenhag, 17-43.
14. Bodrumlu E and Alaçam T. (2006): Evaluation of antimicrobial and antifungal effects of iodoform-integrating gutta-percha. *Cand Dent Assoc* 72(8):733-733d.
15. Webb B, Thomas C, Wilcox M, Harty D, and Knox K.(1998) : *Candida* associated denture stomatitis, Aetiology and management, A review, part 2 oral disease caused by *Candida* species. *Australian Dental Journal* 43 (3):160 – 66.
16. Milne L, Fungi I, Collee J, Fraser A, Marmian B and Simmons A. (1996): *Practical medical microbiology*. 4th ed. 13y Pearson professional limited 695-754.
17. Heelan JS, Sotomayor E, Coon K and Arezzo JB.(1998): Comparison of the Rapid Yeast Plus Panel with the API20C Yeast System for Identification of Clinically Significant Isolates of *Candida* Species. *Journal Clinical Microbiology* 36 (5): 1443-1445.[IVSL high wire].

18. Nolte WA. (1982): Oral microbiology, with basic microbiology and immunology. 4th ed. The C.V. Mosby Company: 287-326.
19. Holbrook W, and Beighton D. (1986): Streptococcus mutans levels in saliva and distribution of Serotypes among 9 years old Icelandic children. Scan Dent Res 95(1):37-42.
20. Perez C., Pauli M. and Bazevque P. (1990): An antibiotic assay by the agar well diffusion method. Acta Biologicae et Medicine Experimentalis 15, 113-115.
21. Eman M. El-Diasty, Ahmed, M.A, Okasha,N, and Salwa, F. Mansour (2013): Antifungal activity of ZnO nanoparticles 23(3): 191–202.
22. Shi LE, Liangying X, Baochao H, Hongjuan G, Xiaofeng G. and Zhenxing T. (2010): Inorganic nano mental oxides used as anti-microorganism agents for pathogen control, in: Current Research, Technology and Education Topics in Applied Microbiology and Microbial Biotechnology 1 :361–368.
23. Lipovsky AY, Nitzan A, Gedanken and R. Lubart,(2011): Antifungal activity of ZnO nanoparticles-the role of ROS mediated cell injury, Nanotechnology. 22:105–101.
24. Jehad M, Yousef and Enas N. Danial.(2012): In Vitro Antibacterial Activity and Minimum Inhibitory Concentration of Zinc Oxide and Nano-particle Zinc oxide Against Pathogenic Strains Journal of Health Sciences 2(4): 38-42.